Using Thermal Energy

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Using Thermal Energy

Lesson 18.3: True or False			
Name_	Class Date		
Detern	nine if the following statements are true or false.		
	1. The function of a thermostat is to transfer thermal energy.		
	2. The water in a hot-water heating system is heated by a furnace.		
	3. In a warm-air heating system, pipes carry thermal energy throughout the house.		
	4. Thermal energy from inside a refrigerator changes the refrigerant to a gas.		
	5. A combustion engine burns fuel to produce thermal energy.		
	6. In any combustion engine, the engine does the work of moving a piston.		
	7. In a warm-air heating system, warm-air vents are always placed near the ceiling.		
	8. An air conditioner is an example of a cooling system.		
	9. Refrigerant changes to a liquid in the condenser of a refrigerator.		
	10. Steam ships have internal combustion engines.		
Less	on 18.3: Critical Reading		
Name_	Class Date		

Cooling Systems

Cooling systems, such as air conditioners and refrigerators, transfer thermal energy in order to keep homes and cars cool or to keep food cold. In a refrigerator, for example, thermal energy is transferred from the cool air inside the refrigerator to the warmer air in the kitchen. Thermal energy naturally moves from a warmer area to a cooler area, so how can it move from the cooler refrigerator to the warmer room? The answer is work. The refrigerator does work to transfer thermal energy in this way. Doing this work takes energy, which is usually provided by electricity.

The key to how a refrigerator (or other cooling system) works is the refrigerant. A refrigerant is a substance, such as FreonTM, that has a low boiling point and changes back and forth between liquid and gaseous states as it cycles through the refrigerator. As a liquid, the refrigerant absorbs thermal energy from the cool air inside the refrigerator. The thermal energy causes the refrigerant to change to a gas. As a gas, the refrigerant releases thermal energy to the warm air outside the refrigerator. This causes the refrigerant to change back to a liquid, and the cycle repeats.

Questions

- 1. What is a cooling system?
- 2. What work must a cooling system do?
- 3. What is the role of the refrigerant in a cooling system?

Read this passage from the text and answer the questions that follow.

Lesson	18.3:	Multip	le Cl	hoice
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Name	Class	Date	_
Circle the letter of the	e correct choice.		

- 1. Types of home heating systems include
 - a. warm-air heating systems.
 - b. hot-water heating systems.
 - c. solar heating systems.
 - d. all of the above
- 2. How is thermal energy transferred in a refrigerator?
 - a. from the warm kitchen to the cool refrigerator
 - b. from the cool refrigerator to the warm kitchen
 - c. from the cool refrigerator to the cold outdoors
 - d. two of the above
- 3. Why must a cooling system do work to keep things cool?
 - a. It transfers thermal energy from a cooler to a warmer place.
 - b. It takes energy to reverse the normal direction of heat flow.
 - c. It takes energy to maintain the normal direction of heat flow.
 - d. two of the above
- 4. What happens to the refrigerant as it passes through a cooling system?
 - a. It freezes and lowers the temperature of the system.
 - b. It changes between liquid and gaseous states.
 - c. It releases thermal energy into the refrigerator.
 - d. It keeps evaporating and has to be replaced.
- 5. In an external combustion engine, thermal energy is used directly to
 - a. move the piston back and forth.
 - b. move the piston up and down.
 - c. turn water into steam.
 - d. all of the above
- 6. What happens first in an internal combustion engine?
 - a. Exhaust gases exit the cylinder.
 - b. The piston moves up or down.
 - c. A fuel-air mixture enters the cylinder.
 - d. The piston rod turns the crankshaft.
- 7. Thermal energy from a radiator travels throughout the air in a room by
 - a. conduction.
 - b. convection.
 - c. radiation.
 - d. all of the above

Lesson	18.3: N	latching
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Name		Date
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www.ck12.org			Chapter 1. Using Thermal Energy
Match each defini	tion with the correct term.		
Definitions			
1. substanc	e that absorbs and releases	thermal energy in a cooling	system
2. device in	n a heating system that cont	rols the furnace or boiler	
3. complex	x machine that produces th	nermal energy outside the m	achine and uses the thermal energy to do
work			
4. heating s	system that includes a boile	er, pipes, and radiators	
5. complex	machine that produces there	rmal energy inside the machi	ine and uses the thermal energy to do work
6. refrigera	tor or air conditioner		
7. heating s	system that includes a furna	ace, ducts, and vents	
Terms			
a. internal combus	stion engine		
b. cooling system			
c. refrigerant			
d. warm-air heatin	ng system		
e. external combu			
f. hot-water heating	-		
g. thermostat	8 - 7		
8			
Lesson 18.3:	Fill in the Blank		
Name	Class	Date	
Fill in the blank w	vith the appropriate term.		
1. As hot wate	er flows through the pipes at	nd radiators of a hot-water ho	eating system, the water becomes
·	r nows unough the pipes as	nd radiators of a not water in	butting system, the water seconds
2. Vents are pl	aced near the floor in a war	m-air heating system becaus	se warm air is low in density and
3. A cooling s	vstem transfers thermal ene	ergy from a cooler area to a v	varmer area by doing
· .		••	oiling point called a(n)
5. Any comple	ex machine that burns fuel	to produce thermal energy as	nd then uses the energy to do work is a(n)
6 The type of	engine that is found in mos	st motor vehicles is a(n)	combustion engine.
	_	ed a(n) combus	
Lesson 19 2:	Critical Writing		
Lesson 10.3.	Cittical Willing		

Name______ Class_____ Date_____

Thoroughly answer the question below. Use appropriate academic vocabulary and clear and complete sentences.

Explain how an external combustion engine produces thermal energy and uses it to do work.